## 5.3 Analytical pyrolysis

This section defines terms and definitions used in analytical methods of pyrolysis and includes expressions for coupled systems and for the description of the temperature profiles and the products that are obtained.

#### **Analytical Pyrolysis**

The characterization, in an inert atmosphere, of a material or a chemical process by a chemical degradation reaction(s) induced by thermal energy.

#### **Catalytic Pyrolysis**

Pyrolysis that is influenced by the addition of a catalyst.

#### Char

A solid carbonaceous pyrolysis residue.

#### **Coil Pyrolyser**

A pyrolyser in which the sample (sometimes located in a tubular vessel) is placed in a metal coil that is heated to cause pyrolysis.

## **Continuous Mode (furnace) Pyrolyser**

A pyrolyser in which the sample is introduced into a furnace preheated to the final temperature.

#### **Curie-point Pyrolyser**

A pyrolyser in which a ferromagnetic sample carrier is inductively heated to its Curie point.

#### Filament (ribbon) Pyrolyser

A pyrolyser in which the sample is placed on a metal filament (ribbon) that is resistively heated to cause pyrolysis.

## Final Pyrolysis Temperature (T<sub>(f, Py)</sub>)

The final (steady state) temperature which is attained by a pyrolyser. (The terms "equilibrium temperature" and "pyrolysis temperature" may be used when referring to an isothermal pyrolysis; they are not recommended for use with a non-isothermal pyrolysis.)

## Flash Pyrolysis

A pyrolysis that is carried out with a fast rate of temperature increase, of the order of 10000 K/s.

#### **Fractionated Pyrolysis**

A pyrolysis in which the same sample is pyrolysed at different temperatures for different times in order to study special fractions of the sample.

#### **In-source Pyrolysis**

A pyrolysis in which the reactor is located within the ion source of a mass spectrometer.

## **IR-pyrogram**

Chromatogram of a pyrolysate detected by infrared spectrometry.

## **Isothermal Pyrolysis**

A pyrolysis during which the temperature is essentially constant.

## Maximum Pyrolysis Temperature (T(max, Py))

The highest temperature in a temperature/time profile

## MS-pyrogram

Chromatogram of a pyrolysate detected by mass spectrometry.

# **Off-line Pyrolysis**

A pyrolysis in which the products are trapped before analysis.

## **Oxidative Pyrolysis**

A pyrolysis that occurs in the presence of an oxidative atmosphere.

## **Pressure Monitored Pyrolysis**

A pyrolysis technique in which the pressure of the volatile pyrolysates is recorded as the sample is heated.

#### **Pulse Mode Pyrolyser**

A pyrolyser in which the sample is introduced into a cold furnace which is then heated rapidly.

## Pyrogram

A chromatogram of a pyrolysate.

## **Pyrolysate (pyrolyzate)**

The products of pyrolysis.

## Pyrolyser (pyrolyzer)

A device for performing pyrolysis.

## **Pyrolysis**

A chemical degradation reaction that is caused by thermal energy. (The term *pyrolysis* generally refers to an inert environment.)

# Pyrolysis-gas Chromatography (Py-GC)

A pyrolysis technique in which the volatile pyrolysates are directly conducted into a gas

chromatograph for separation and detection.

# Pyrolysis-gas Chromatograph-mass Spectrometry (Py-GC-MS)

A pyrolysis technique in which the volatile pyrolysates are separated and analyzed by online gas chromatography-mass spectrometry.

# Pyrolysis-Gas Chromatography-Infrared Spectroscopy (Py-GC-IR)

A pyrolysis technique in which the volatile pyrolysates are separated and analyzed by online gas chromatography-infrared spectroscopy.

# Pyrolysis-Infrared Spectroscopy (Py-IR)

A pyrolysis technique in which the pyrolysates are detected and analyzed by on-line infrared spectroscopy.

# Pyrolysis-Infrared Spectrum

Infrared spectrum obtained from *pyrolysis-infrared spectroscopy*.

# Pyrolysis-Mass Spectrometry (Py-MS)

A pyrolysis technique in which the volatile pyrolysates are detected and analyzed by on-line mass spectrometry.

# **Pyrolysis-Mass Spectrum**

Mass spectrum obtained from pyrolysis-mass spectrometry.

# **Pyrolysis Reactor**

That portion of the pyrolyser in which the pyrolysis takes place.

# **Pyrolysis Residue**

That portion of the pyrolysate that does not leave the reactor.

# **Pyrolysis Thermogram**

The result of a temperature programmed pyrolysis in which the detector signals e.g. total ion current or single ions, total absorbance or a GC-detector, are plotted against time or temperature.

## **Reductive Pyrolysis**

A pyrolysis which occurs in the presence of a reducing atmosphere.

## **Sequential Pyrolysis**

A pyrolysis in which the same initial sample is repetitively pyrolysed under identical conditions (final pyrolysis temperature, temperature rise time and total heating time).

#### **Stepwise Pyrolysis**

A pyrolysis in which the sample temperature is raised stepwise. The pyrolysis products are recorded between each step.

#### Tar

A liquid pyrolysis residue.

## **Temperature-Programmed Pyrolysis**

A pyrolysis during which the sample is heated at a controlled rate within a temperature range in which pyrolysis occurs.

#### **Temperature Rise Time (TRT)**

The time required for a pyrolyser temperature to be increased from its initial to its final temperature.

## **Temperature Time Profile (TTP)**

A graphical representation of temperature *versus* time for a particular pyrolysis experiment or pyrolyser.

# **Total Heating Time (THT)**

The time between the onset and conclusion of the sample heating in a pyrolysis experiment.

# Volatile Pyrolyzate

That portion of the pyrolysate which has adequate vapour pressure to reach the detector.